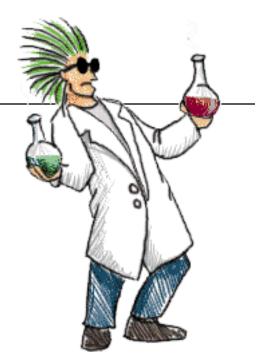
Physical and Chemical Changes

A Write On Activity



Objective: Chemical/Physical Change: Differentiate

The learner will be able to (ESSENTIAL) distinguish between a physical and chemical change.



Matter

- □ Matter is everywhere.
- Matter is anything that takes up space and has mass.
- Matter is constantly experiencing both chemical and physical changes.

- Physical changes occur when matter changes its property but not its chemical nature.
- Physical property changes could include a change in: texture, shape, size, color, odor, volume, mass, weight, and density.







Chemical Change

- Chemical changes are changes matter undergoes when it becomes new or different matter.
- To identify a chemical change look for signs such as color change, bubbling and fizzing, light production, smoke, and presence of heat.

Chemical Change

A chemical change occurs when fireworks are used. Fireworks are made of metals such as magnesium and copper. These change chemically as they light up the sky.



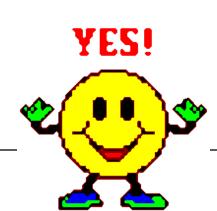
Sugar dissolving in tea

Chemical Change











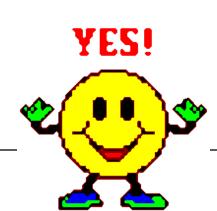
Logs burning

Chemical Change





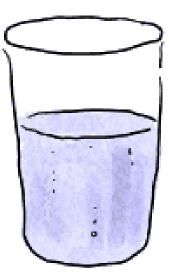






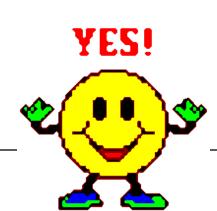
Breaking water up by separating it into hydrogen and oxygen

Chemical Change









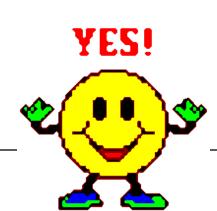


Cutting paper

Chemical Change









Crushing an aspirin

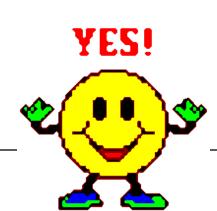
Chemical Change













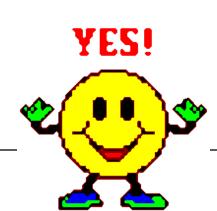
Metal rusting

Chemical Change











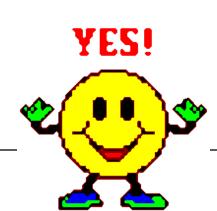
Lighter fluid burning

Chemical Change











An egg rotting

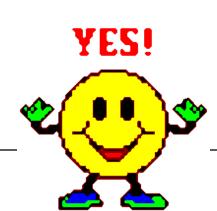
Chemical Change

Smells Like...









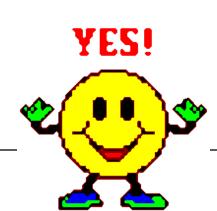


An egg breaking

Chemical Change









Writing Activity for your ISN

- Write a paragraph about the difference between a chemical and physical change.
- Give examples of each.

